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Navarrini et al.

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[54] PERFLUORODIAZIRIDINES AND
PROCESS FOR PREPARING THEM

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[30] Foreign Application Priority Data

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[51] Int. Cl.⁵ C07D 229/02

[52] U.S. Cl. 548/960

[58] Field of Search 548/960

[56] References Cited

U.S. PATENT DOCUMENTS

3,345,360 10/1967 Firth 548/960

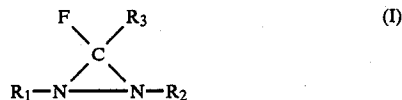
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[57] ABSTRACT

Novel perfluorodiaziridines having the formula:



wherein:

R¹ and R₂, alike or different from each other, represent a perfluoroalkyl group of from 1 to 10 carbon atoms, and

R₃ represents a fluorine atom, or a perfluoroalkyl group of from 1 to 9 carbon atoms.

These perfluorodiaziridines are obtained by reacting perfluoroaminooxaziridines with a source of fluoride ions.

2 Claims, No Drawings

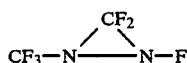
PERFLUORODIAZIRIDINES AND PROCESS FOR PREPARING THEM

DESCRIPTION OF THE INVENTION

The present inventions relates to novel perfluorodiaziridines and to a process for preparing them.

Perfluorodiaziridines are a not very widely known class of organic compounds.

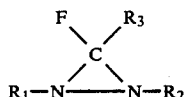
The perfluorodiaziridine of formula:



is known. This was synthesized by starting from $\text{CF}_2=\text{N}-\text{F}$ and CsF (Shi-Ching Chang and Darryl D. DesMarteau, *J. Org. Chem.* 1983, 48, 771-774).

One purpose of the present invention is to provide a novel class of perfluorodiaziridines. A further purpose is to provide a process for preparing them.

The first purpose is achieved by providing the novel perfluorodiaziridines according to the present invention having the formula:



wherein:

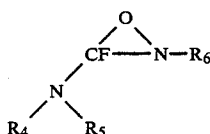
R_1 and R_2 , alike or different from each other, represent a perfluoroalkyl group of from 1 to 10 carbon atoms, and

R_3 represents a fluorine atom, or a perfluoroalkyl group of from 1 to 9 carbon atoms.

These novel perfluorodiaziridines are useful as catalysts for the photochemical polymerization of olefinic monomers. They form complexes with transition metal ions, and are useful as intermediates in the preparation of nitrenes.

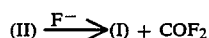
Preferably, R_1 and R_2 are perfluoroalkyl groups of from 1 to 3 carbon atoms, and R_3 is a fluorine atom, or a perfluoroalkyl group of from 1 to 2 carbon atoms.

These new perfluorodiaziridines may be prepared by reacting a perfluoroamino-oxaziridine of formula:



wherein: R_4 , R_5 and R_6 , alike or different from one another, are perfluoroalkyl groups containing from 1 to 10 carbon atoms, with a source of fluoride ions.

The reaction may be schematically shown as follows:



The R_3 radical of (I) derives from the R_4 or R_5 radical of (II), and contains one carbon atom less than this latter, so that R_3 is F when its source radical is $-\text{CF}_3$.

The R_1 and R_2 radicals of (I) derive from radicals of (II) different from the radical which supplied the R_3 .

The reaction is commonly carried out at a temperature within the range of from 0° to 120° C.

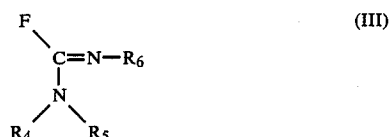
As the fluoride ion sources, in particular, CsF , KF and tetraalkyl-ammonium fluorides are used.

The molar ratio of the fluoride ion source, computed as CsF , to the perfluoroaminooxaziridines (II) is generally within the range of from 0.1 to 10, and preferably from 1 to 10.

The reaction may be carried out in the presence of a dipolar aprotic solvent, such as acetonitrile, the glymes, dimethylformamide, and dimethylsulphoxide.

The perfluoroaminooxaziridines (II) and the process for preparing them are disclosed in U.S. Pat. No. 4,874,875, which is incorporated herein by reference.

According to that patent, perfluoroaminooxaziridines (II) are obtained by reacting a perfluoroimine of formula:

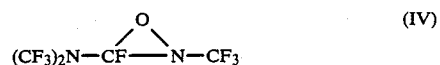


wherein R_4 , R_5 and R_6 have the above meanings, with H_2O_2 in the presence of a base, in a dipolar aprotic solvent, at a temperature within the range of from -50° to $+50^\circ$ C.

The following example is provided as illustrative and not limitative of the present invention.

EXAMPLE

To a 150 ml glass reactor containing 3 g of CsF , 5 mmol is charged of a perfluoroaminooxaziridine of formula:



The so-charged reactor is maintained at room temperature for 8 hours.

The raw reaction product is distilled under a pressure of 10^{-3} torr.

The vapors from the distillation kettle flow through cold traps maintained at a temperature of -120° C. and of -196° C., respectively.

Inside the trap at -120° C. 3 mmol condenses of 1,2-trifluoromethyl-3,3-difluorodiaziridine having the formula:



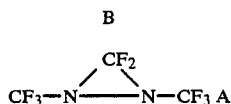
with a yield of 60% relative to the perfluoroaminooxaziridine used as the starting material. Inside the trap a mixture condenses at -196° C. which is prevalingly constituted by COF_2 , together with byproducts.

The diaziridine (V) was analyzed by I.R. spectrum, ^{19}F N.M.R., and mass spectrum.

The main absorption bands in the I.R. range are the following: cm^{-1} (intensity): 1443 (s), 1317 (vs), 1277 (s), 1245 (vs), 1205 (s) and 996 (m), wherein "vs" stands for "very strong", "s" stands for "strong" and "m" stands for "medium."

The N.M.R. spectrum (internal reference CFCl_3 ; solvent CdCl_3), gave:

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A=(triplet) - 65.6 ppm JAB 8 Hz

B=(heptet) - 108.6 ppm JAB 8 Hz

The mass spectrum gave:

M, 216 (1.8%); 69 (100%); 128 (31.3%); 197 (19.1%).

Although the invention has been described in conjunction with specific embodiments, it is evident that many alternatives and variations will be apparent to those skilled in the art in light of the foregoing description. Accordingly, the invention is intended to embrace all of the alternatives and variations that fall within the spirit and scope of the appended claims.

What is claimed is:

1. A perfluorodiaziridine having the formula:

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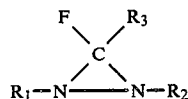
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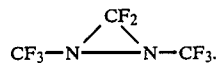


wherein:

R₁ and R₂, alike or different from each other, represent a perfluoroalkyl group of from 1 to 3 carbon atoms, and

R₃ represents a fluorine atom, or a perfluoroalkyl group of from 1 to 2 carbon atoms.

2. 1,2-trifluoromethyl-3,3-difluorodiaziridine having the formula:



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